

Bound State Solutions of the Schrodinger's Equation with Manning-Rosen plus Yukawa Potential using Pekeris-like Approximation of the Coulomb Term and Parametric Nikoforov-Uvarov

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ABSTRACT

The solutions of the Schrödinger equation with Manning-Rosen plus Yukawa potential (MRYP) have been presented using the Pekeris-like approximation of the coulomb term and parametric Nikiforov-Uvarov (NU) method. The bound state energy eigenvalues and the corresponding un-normalized eigen functions are obtained in terms of Jacobi polynomials. Also, Yukawa, Manning-Rosen and coulomb potentials have been recovered from the mixed potential and their eigen values obtained. The Numerical results are computed for some values of n at $l=0$ with $\alpha = 0.01, 0.1, 2$ and 5 using python 3.6 programming, and these results could be applied to molecules moving under the influence of MRYP potential as negative energy eigenvalues obtained indicate a bound state system.

Keywords: Schrödinger equation, Manning-Rosen potential, Yukawa potential, Pekeris-like approximation, Parametric Nikiforov-Uvarov method, Jacobi polynomials

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1 INTRODUCTION

In quantum mechanics, one of the interesting problems is to obtain exact solutions of the Schrödinger equation. In order to do this, a real potential is normally chosen to derive the energy eigenvalues and the eigen functions of the Schrödinger equation.¹ These solutions describe the particle dynamics in non-relativistic quantum mechanics.² Several authors have studied the bound states of the Schrödinger equation using different potentials and methods. Some of these potentials play very important roles in many fields of Physics such as Molecular Physics, Solid State

and Chemical Physics.³ The Manning-Rosen potential has been deeply studied and applied in quantum systems and Yukawa potential and its classes have been studied in Schrodinger formalism.⁴

The purpose of the present paper is to solve the Schrödinger equation for the mixed potential MRYP using the parametric NU method. The paper is organized as follows: After a brief introduction in section 1, the NU method is reviewed in section 2. In section 3, we solve the radial Schrödinger equation using the NU method. Finally, we discuss our results in section

4 and a brief conclusion is then advanced in section 5.

2 NIKIFOROV-UVAROV METHOD

The Nikiforov-Uvarov (NU) method⁵ is based on the solutions of a generalized second-order linear differential equation with special orthogonal functions. The Schrödinger equation and Schrödinger-like equations of the type as:

$$\psi''(r) + [E - V(r)]\psi(r) = 0, \tag{1}$$

can be solved by this method. To do this equation (1) is transformed into equation of hypergeometric type with appropriate coordinate transformation $s = s(r)$ to get

$$\psi''(s) + \frac{\bar{\tau}(s)}{\sigma(s)}\psi'(s) + \frac{\bar{\sigma}(s)}{\sigma^2(s)}\psi(s) = 0, \tag{2}$$

To solve equation (2) we can use the parametric NU method. The parametric generalization of the NU method is expressed by the generalized hypergeometric type equation^[19]

$$\psi''(s) + \frac{(c_1 - c_2s)}{s(1 - c_3s)}\psi'(s) + \frac{1}{s^2(1 - c_3s)^2}[-\epsilon_1s^2 + \epsilon_2s - \epsilon_3]\psi(s) = 0, \tag{3}$$

where $\sigma(s)$ and $\bar{\sigma}(s)$ are polynomials at most second degree, and $\bar{\tau}(s)$ is a first degree polynomial. The eigenfunctions (equation 4) and corresponding eigenvalues (equation 5) to the equation become

$$\psi(s) = N_n s^{c_{12}} (1 - c_3s)^{-c_{12} - \frac{c_{13}}{c_3}} P_n^{(c_{10}-1, \frac{c_{11}}{c_3} - c_{10}-1)}(1 - 2c_3s), \tag{4}$$

$$(c_2 - c_3)n + c_3n^2 - (2n + 1)c_5 + (2n + 1)(\sqrt{c_9} + c_3\sqrt{c_8}) + c_7 + 2c_3c_8 + 2\sqrt{c_8c_9} = 0, \tag{5}$$

Where

$$c_4 = \frac{1}{2}(1 - c_1), c_5 = \frac{1}{2}(c_2 - 2c_3), c_6 = c_5^2 + \epsilon_1, c_7 = 2c_4c_5 - \epsilon_2, c_8 = c_4^2 + \epsilon_3, c_9 = c_3c_7 + c_2^2c_8 + c_6, c_{10} = c_1 + 2c_4 + 2\sqrt{c_8}, c_{11} = c_2 - 2c_5 + 2(\sqrt{c_9} + c_3\sqrt{c_8}), c_{12} = c_4 + \sqrt{c_8}, c_{13} = c_5 - (\sqrt{c_9} + c_3\sqrt{c_8}), \tag{6}$$

N_n is the normalization constant and $P_n^{(\alpha, \beta)}$ are the Jacobi polynomials.

3. SOLUTIONS OF THE RADIAL PART OF SCHRÖDINGER EQUATION WITH MRYP POTENTIAL

The radial Schrödinger equation⁶ is given as

$$\frac{d^2R_{nl}(r)}{dr^2} + \frac{2\mu}{\hbar^2} \left[E - V(r) - \frac{\lambda\hbar^2}{2\mu r^2} \right] R_{nl}(r), \tag{7}$$

Where $\lambda = l(l + 1)$ and $V(r)$ is the potential energy function. The Manning-Rosen potential (MRP) is given as:

$$V(r) = - \left[\frac{Ce^{-\alpha r} + De^{-2\alpha r}}{(1 - e^{-\alpha r})^2} \right] \tag{8}$$

The Yukawa potential (YP) is given as:

$$V(r) = - \frac{V_0 e^{-\alpha r}}{r}, \tag{9}$$

where V_0 is the potential depth of the YP and α is an adjustable positive parameter. In equation (8) C and D are constants. The sum of these potentials known as MRYP is given as

$$V(r) = - \left[\frac{Ce^{-\alpha r} + De^{-2\alpha r}}{(1 - e^{-\alpha r})^2} \right] - \frac{V_0 e^{-\alpha r}}{r} \tag{10}$$

Making the transformation $s = e^{-\alpha r}$ equation (10) becomes

$$V(s) = - \left[\frac{CS + DS^2}{(1 - s)^2} \right] - \frac{\alpha V_0 s}{1 - s} \tag{11}$$

Again, applying the transformation $s = e^{-\alpha r}$ and using the Pekeris-type approximation⁷, we get the form that NU method is applicable, equation (7) gives a generalized hypergeometric-type equation as

$$\frac{d^2R(s)}{ds^2} + \frac{(1-s)}{(1-s)s} \frac{dR(s)}{ds} + \frac{1}{(1-s)^2 s^2} [-(\beta^2 - F + B)s^2 + (2\beta^2 + A + B)s - (\beta^2)]R(s) = 0, \tag{12}$$

Where $\lambda = 0, -\beta^2 = \frac{2\mu E}{\alpha^2 \hbar^2}, A = \frac{2\mu C}{\alpha^2 \hbar^2}, B = \frac{2\mu V_0}{\alpha \hbar^2}, F = \frac{2\mu D}{\alpha^2 \hbar^2}, \frac{1}{r} \approx \frac{\alpha}{(1-e^{-\alpha r})} \approx \frac{\alpha}{(1-s)},$ (13)

Comparing equation (12) with equation (3) yields the following parameters

$$c_1 = c_2 = c_3 = 1, c_4 = 0, c_5 = -\frac{1}{2}, c_6 = \frac{1}{4} + \beta^2 + B - F, c_7 = -2\beta^2 - A - B, c_8 = \beta^2, c_9 = \frac{1}{4} - (A + F), c_{10} = 1 + 2\sqrt{\beta^2}, c_{11} = 2 + 2\left(\sqrt{\frac{1}{4} - A - F} + \sqrt{\beta^2}\right), c_{12} = \sqrt{\beta^2}, c_{13} = -\frac{1}{2} - \left(\sqrt{\frac{1}{4} - A - F} + \sqrt{\beta^2}\right), \epsilon_1 = \beta^2 + B - F, \epsilon_2 = 2\beta^2 + A + B, \epsilon_3 = \beta^2,$$
 (14)

Now using equations (5), (13) and (14) we obtain the energy eigen spectrum of the MRYP as

$$\beta^2 = \left[\frac{A+B - (n^2+n+\frac{1}{2}) - (2n+1)\sqrt{\frac{1}{4}-A-F}}{(2n+1)+2\sqrt{\frac{1}{4}-A-F}} \right]^2,$$
 (15)

Equation (15) can be solved explicitly and the energy eigen spectrum of MRYP becomes

$$E = -\frac{\alpha^2 \hbar^2}{2\mu} \left\{ \left[\frac{\frac{2\mu C}{\alpha^2 \hbar^2} + \frac{2\mu V_0}{\alpha \hbar^2} - (n^2+n+\frac{1}{2}) - (2n+1)\sqrt{\frac{1}{4} - \frac{2\mu C}{\alpha^2 \hbar^2} - \frac{2\mu D}{\alpha^2 \hbar^2}}}{(2n+1)+2\sqrt{\frac{1}{4} - \frac{2\mu C}{\alpha^2 \hbar^2} - \frac{2\mu D}{\alpha^2 \hbar^2}}} \right]^2 \right\}$$
 (16)

We now calculate the radial wave function of the MRYP as follows

The weight function $\rho(s)$ is given as [19]

$$\rho(s) = s^{c_{10}-1} (1 - c_3 s)^{\frac{c_{11}}{c_3} - c_{10} - 1},$$
 (17)

Using equation (14) we get the weight function as

$$\rho(s) = s^U (1 - s)^V,$$
 (18)

Where $U = 2\sqrt{\beta^2}$ and $V = 2\sqrt{\frac{1}{4} - A - F}$

Also we obtain the wave function $\chi(s)$ as [19]

$$\chi(s) = P_n^{c_{10}-1, \frac{c_{11}}{c_3} - c_{10} - 1} (1 - 2c_3 s),$$
 (19)

Using equation (14) we get the function $\chi(s)$ as

$$\chi(s) = P_n^{(U,V)} (1 - 2s),$$
 (20)

Where $P_n^{(U,V)}$ are Jacobi polynomials

Lastly,

$$\varphi(s) = s^{c_{12}} (1 - c_3 s)^{-c_{12} - \frac{c_{13}}{c_3}},$$
 (21)

And using equation (14) we get

$$\varphi(s) = s^{U/2} (1 - s)^{V-1/2},$$
 (22)

We then obtain the radial wave function from the equation [19]

$$R_n(s) = N_n \varphi(s) \chi_n(s),$$
 (23)

As

$$R_n(s) = N_n s^{U/2} (1 - s)^{(V-1)/2} P_n^{(U,V)} (1 - 2s),$$
 (24)

Where n is a positive integer and N_n is the normalization constant.

4 DISCUSSION:

We have solved the radial Schrödinger equation and obtained the energy eigen values for the Manning-Rosen plus Yukawa potential (MRYP) in equation (16). We calculated the bound state energy eigenvalues by variation of the screening parameter as shown in Table 1.

Table 1. Energy eigenvalues $E(eV)$ of the MRYP potential for $\hbar=\mu=1, V_0 = 0.2, C = -0.1, D = 0.1$ with different α values.

n	$\alpha = 0.01$	$\alpha = 0.1$	$\alpha = 2$	$\alpha = 5$
1	-12.0540	-0.12499	-1.85281	-12.05405
2	-5.3846	-0.08680	-4.35124	-27.67680
3	-3.05045	-0.08	-7.85070	-49.55101
4	-1.9701	-0.08405	-12.3504	-77.67564
5	-1.38333	-0.09388	-17.8503	-112.0504
6	-1.02961	-0.10778	-24.3502	-152.6753
7	-0.80011	-0.12500	-31.8501	-199.5502

The following cases are considered:

Case 1: If $C = D = 0$ in equation (10), the potential turns back into the Yukawa potential and equation (16) yields the energy eigen values of the Yukawa potential as

$$E = -\frac{\hbar^2}{2\mu} \left[\frac{\frac{2\mu V_0}{\hbar^2} - \alpha^2(n+1)^2}{2(n+1)} \right]^2, \quad (25)$$

Equation (25) is similar to equation (30) obtained by Antia *et al.*, 2015

Case 2: If $\alpha \rightarrow 0, V_0 = Ze^2$ in equation (25), the energy eigen values for Coulomb potential becomes

$$E = -\frac{Z^2 e^4 \mu}{2\hbar^2 n'^2} \quad (26)$$

Where $n' = n + 1$ in this case.

Case 3: If $V_0 = 0$ the potential in equation (10) yields the Manning-Rosen potential with energy eigen values given as

Eq. (27) is also similar to Manning-Rosen potential bound state energy obtained by Louis *et al.*⁴

$$E = -\frac{\alpha^2 \hbar^2}{2\mu} \left\{ \frac{\left[\frac{2\mu C}{\alpha^2 \hbar^2} - (n^2 + n + \frac{1}{2}) - (2n+1) \sqrt{\frac{1}{4} \frac{2\mu C}{\alpha^2 \hbar^2} \frac{2\mu D}{\alpha^2 \hbar^2}} \right]^2}{(2n+1) + 2 \sqrt{\frac{1}{4} \frac{2\mu C}{\alpha^2 \hbar^2} \frac{2\mu D}{\alpha^2 \hbar^2}}} \right\} \quad (27)$$

5. CONCLUSION:

We have obtained the energy eigen values and the corresponding un-normalized wave function using the parametric NU method for the Schrödinger equation with MRYP. Special cases of the potential have also been considered.

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